## Salicylaldehyde Derivatives as Building Blocks in the Synthesis of Useful Open Chain and Macrocyclic Crown Compounds

Yehia A. Ibrahim,\* Ahmed H. M. Elwahy and Ashraf A. Abbas

Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

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The conversion of salicyaldehyde and its derivatives into new usful polyethers with bis-terminal functional groups, crown diamides and crown ethers is described.

Much attention has been directed to the supramolecular chemistry of crown compounds owing to their diverse applications in many fields. The introduction of an amide linkage in the polyether ring modifies the binding properties of crown compounds in favour of alkaline earth cations with respect to alkali metal ions.<sup>15–20</sup> Also, some polyether diamides and crown diamides showed interesting applications in cation selective electrodes.<sup>25a–f</sup> Moreover, open chain crown compounds with suitable end groups have interesting applications and some of them were shown to form stable crystalline complexes with various metal cations.<sup>26–29</sup>

In continuation of our recent<sup>20,45</sup> interest in synthesizing macrocyclic crown compounds and their precursors starting from salicylaldehyde and its derivatives, we describe here an easy access towards a number of polyethers with terminal bis-functional groups, crown diamides and crown ethers.



Scheme A

\*To receive any correspondence (*e-mail:* Yehia\_Al@FRCU.EUN.-RG).

The starting bis(chloromethyl) derivatives 7a-f and the naphtho analogue 15 were prepared from salicyladehyde (1) *o*-hydroxybenzyl alcohols 9a-c and 2-hydroxynaphthaldehyde (11) as outlined in Scheme A. Thus, alkylation of the appropriate salicyladehyde derivatives 1, 2a,b with the appropriate dihalo compounds gave the corresponding bis(carbonyl)ethers, 5a-c which underwent NaBH<sub>4</sub> reduction to give the corresponding bis(hydroxymethyl) derivatives 6a,d,b, respectively. The latter were alternatively prepared from the appropriate *o*-hydroxybenzyl alcohol derivatives 9a-c. The bis(hydroxymethyl)naphthyl derivative 14 was similarly prepared from 2-hydroxynaphthaldehyde. The bis(hydroxymethyl)ethers 6a-f, 14 were converted into the corresponding bis(chloromethyl) derivatives 7a-f, 15 upon treatment with SOCl<sub>2</sub> in chloroform.

Interestingly, we found that during attempted crystallization of 7a,h,j, 15 from alcohols, alcoholysis took place and the corresponding bis(alkoxymethyl) derivatives 16c-eand 21 were obtained, respectively. The syntheses of the bis(ethoxymethyl) derivatives 16a,b,f were achieved upon treatment of 7a,b,k with sodium ethoxide in refluxing ethanol after 24 h. This is due to the decreased reactivity caused by the electron withdrawing effect of the arylazo and chloro groups as shown in Scheme B.

Nucleophilic attack on **7a–k** with phenoxides, secondary amines and potassium formate gave the corresponding bis-(functionalized)polyether derivatives **18a–i**, **19a–e**, **20a–d**. Similarly, compound **18i** was converted to the diol **18g** then to the corresponding bis(chloromethyl) derivative **18k** which was reacted with **17d** to give the extended  $\alpha,\omega$ -bis(carbonyl)ether **18l**.

Also, nucleophilic reaction of the appropriate bisphenoxides **25a,b**, **28** with **7a–f** gave the corresponding macrocyclic 23,24-crown-diamides **26a–g** and 23-crown-6 derivatives **29a–c**.

Techniques used: 1H and 13C NMR and MS

Schemes: 5 References: 50

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## **References cited in this synopsis**

- 15 Y. Nakatsuji, H. Kobayasni, M. Okahara and K. Matsuch, Chem. Lett., 1982, 1571.
- 16 (a) J. Petranek and O. Ryba, *Tetrahedron Lett.*, 1977, 4249;
  (b) Anal. Chim. Acta, 1981, **128**, 129; (c) Collect. Czech. Chem. Commun., 1980, **45**, 1567; (d) 1983, **48**, 1944.
- 17 (a) K. Kimura, K. Kumami, S. Kitazawa and T. Shono, J. Chem. Soc., Chem. Commun., 1984, 442; (b) K. Kimura, K. Kumami, S. Kitazawa and T. Shono, Anal. Chem., 1984, 56, 2369.
- 18 M. C. Duriez, T. Pigot, C. Picard, L. Cazaux and P. Tisnes, *Tetrahedron*, 1992, **48**, 4347.
- 19 T. Pigot, M. C. Durierz, C. Picard, L. Cazaux and P. Tisnes, *Tetrahedron*, 1992, **48**, 4359.
- 20 Y. A. Ibrahim and A. M. El Wahy, Synthesis, 1993, 503.
- 25 (a) V. P. Y. Gadzekpo, J. M. Hungerford, A. M. Kadry, Y. A. Ibrahim and G. D. Christian, *Anal. Chem.*, 1985, **57**, 493;



## Scheme B

- (b) V. P. Y. Gadzekpo, J. M. Hungerford, A. M. Kadry, Y. A. Ibrahim, R. Y. Xie and G. D. Christian, *Anal. Chem.*, 1986, **58**, 1948; (c) R. Y. Xie, V. P. Y. Gadzekpo, A. M. Kadry, Y. A. Ibrahim, J. Ruzicka and G. D. Christian, *Anal. Chem. Acta*, 1986, **184**, 259; (d) A. S. Attiyat, Y. A. Ibrahim, A. M. Kadry, R. Y. Xie and G. D. Christian, *Z. Anal. Chem.*, 1987, **329**, 12; (e) A. S. Attiyat, A. M. Kadry, H. R. Hanna, Y. A. Ibrahim and G. D. Christian, *Anal. Sci. (Japan)*, 1990, **6**, 233; (f) A. S. Attiyat, A. M. Kadry, M. A. Badawy, H. R. Hanna, Y. A. Ibrahim and G. D. Christian, *Electroanalysis*, 1990, **2**, 119.
- 26 (a) E. Weber and F. Vogtle, *Tetrahedron Lett.* 1975, 2415;
  (b) W. Rabßhofer, G. Open and F. Vogtle, *Chem. Ber.*, 1978, 111, 419.
- 27 F. Vogtle and H. Sieger, Angew. Chem., Int. Ed. Engl., 1977, 16, 396.
- 28 W. Rabßhofer, W. M. Muller and F. Vogtle, *Chem. Ber.*, 1979, 112, 2095.
- 29 U. Heimann and F. Vogtle, Chem. Ber., 1979, 112, 3034.
- 45 Y. A. Ibrahim, A. M. El Wahy and G. M. M. El-Kareish, J. Chem. Res., 1994, (S) 414; (M) 2321.
- 46 D. Sikhibhushan, J. Chem. Soc., 1926, 1171.